

Charged particles

Observing the force

Direction
of
*conventional **current***

Direction
of
*flow of **positive charge***

Direction
of
*flow of **electrons***

Observing the force

Electron beam tube

to demonstrate the magnetic force on a moving charge

Electron gun

a beam of electrons

Magnets

magnetic field

Observing the force

You can
observe



the effect of

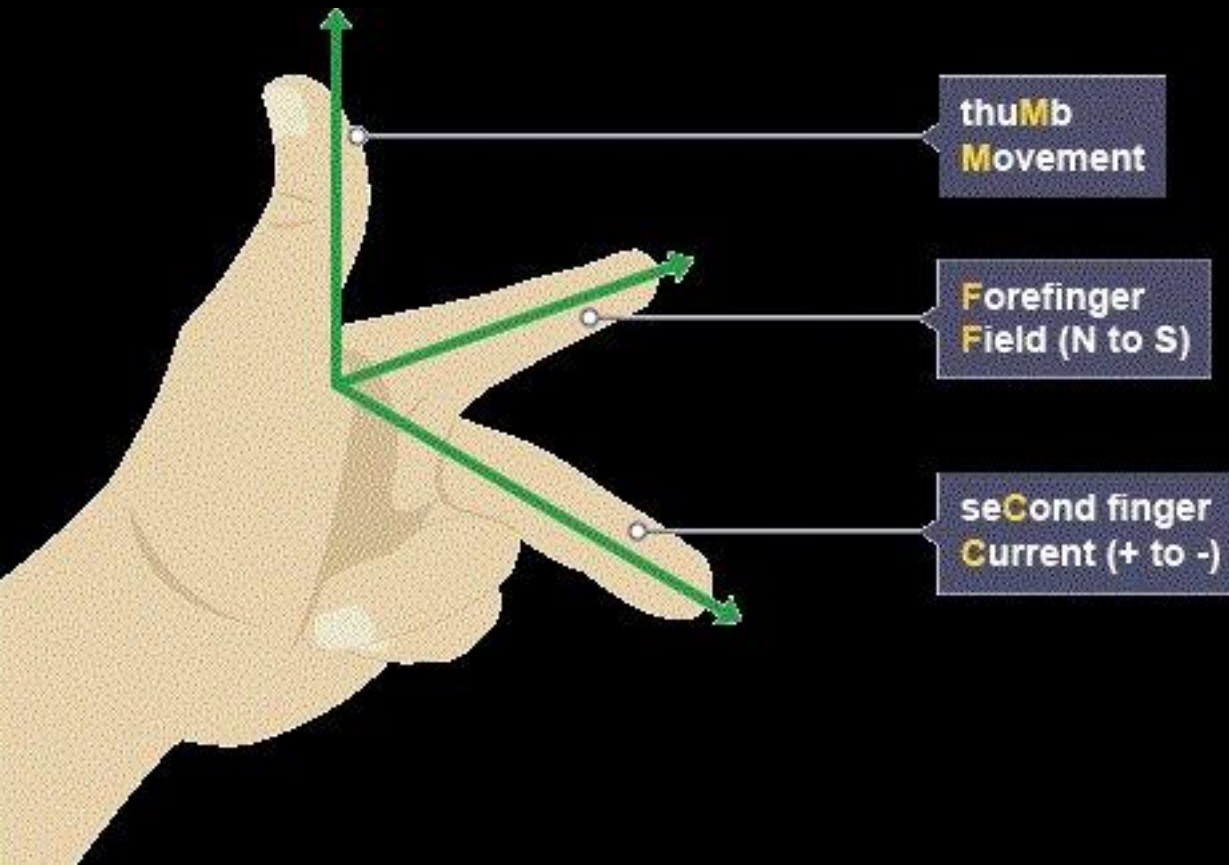
changing the
magnitude of the field

changing the
direction of the field

reversing the field

Observing the force

Fleming's left-hand rule



thumb – force

forefinger – field

second finger – current

Observing the force

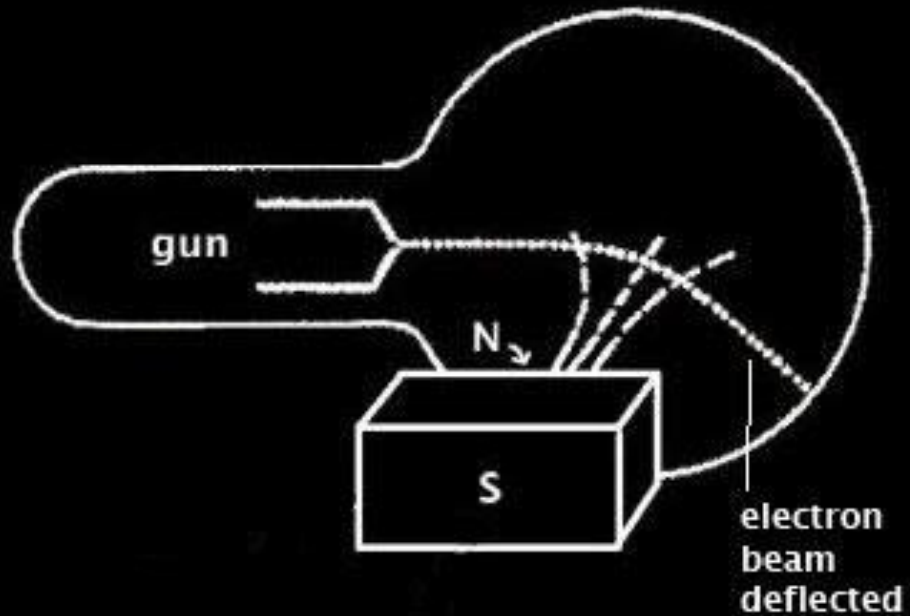
Deflection of electron beams in magnetic field

Electrons – from right to left

Magnetic field – into the plane of paper

Force – upwards

Observing the force



*Force due to the magnetic field is always at right angles to the **velocity** of the electrons*

when direction of the beam changes,
the direction of the force changes

Observing the force

Electron beam tube

Electron gun

a beam of electrons

Magnets

magnetic field

Cathode

electrons

Anode

attracts electrons

Plates

changes direction

The magnetic force on a moving charge

Force on a charge depends on

- the magnetic field flux density B
- the charge Q on the particle
- the speed v of the particle

$$F = BQv$$

'Bev' force

$$F = Bev$$

The magnetic force on a moving charge

Fleming's
left-hand rule

The force is
always at 90° to
the velocity

So arc is formed

The magnetic force on a moving charge

$$F = BIl \quad \longleftrightarrow \quad F = BQv$$

$$I = \frac{Q}{t}$$

The magnetic force on a moving charge

So fucking important reminder:

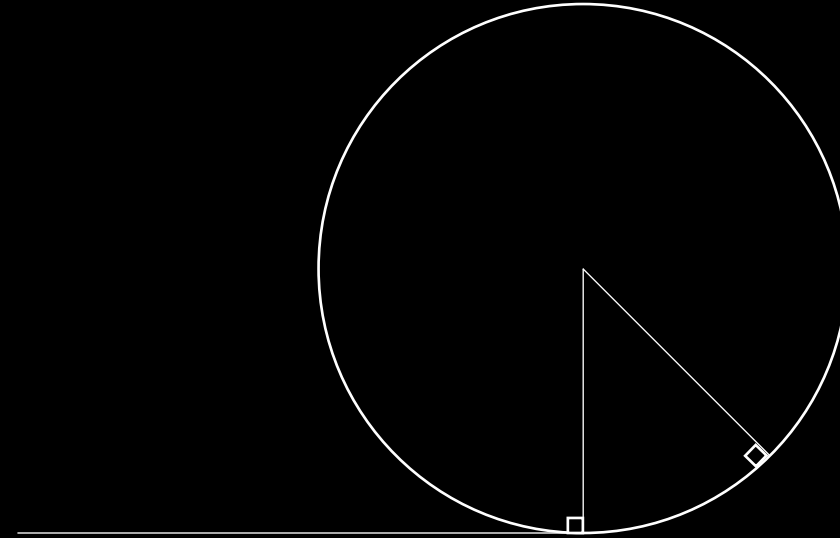
“The **net force** F is always at right-angles to the particle’s **velocity** v ”

Orbiting charges

when a **charged particle** is moving at **right angles** to a uniform **magnetic field**

charged particle forms a **circular** path

because



The magnetic force F is always perpendicular to its velocity v

Orbiting charges

$$\text{centripetal force} = \frac{mv^2}{r}$$

$$\text{centripetal force} = Bev$$

$$r = \frac{mv}{Be}$$

$$P = Ber$$

Orbiting charges

From $r = \frac{mv}{Be}$:

- **Fast** particles form **bigger** circles
- **Heavier** particles form **bigger** circles
- **Weaker** field forms **bigger** circles

$$r \sim v$$

$$r \sim m$$

$$r \sim \frac{1}{B}$$

The charge-to-mass ratio of an electron

To find the mass
of an electron

First find the
charge-to-mass ratio $\frac{e}{m}$

The charge-to-mass ratio of an electron

From $r = \frac{mv}{Be}$:

$$\frac{e}{m_e} = \frac{v}{Br}$$

The charge-to-mass ratio of an electron

To find ν :

cathode-anode voltage V_{ca}

The charge-to-mass ratio of an electron

V_{ca} causes each electron to accelerate

If each electron has charge $-e$

Work done on each electron is $e * V_{ca}$

This is its KE ($= \frac{1}{2} m_e v^2$)

The charge-to-mass ratio of an electron

$$eV_{ca} = \frac{1}{2}m_e v^2 \qquad r = \frac{m_e v}{Be}$$

$$\frac{e}{m_e} = \frac{2V_{ca}}{r^2 B^2}$$

Now, we need [V_{ca} | r | B]

Electric and Magnetic fields

Deflection tube

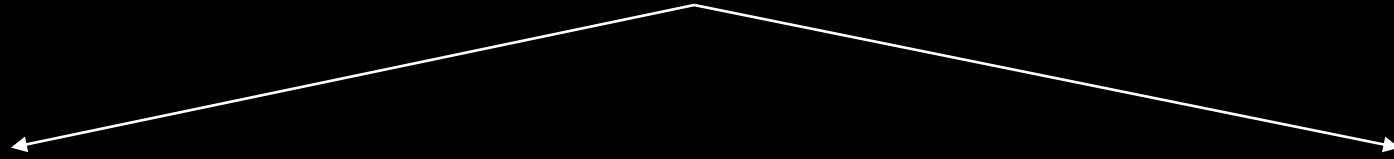
a beam passes through electric and magnetic fields

to keep the beam horizontal, you can balance the forces the magnetic and electric fields provide

Electric and Magnetic fields

To keep the electron beam to be **straight**:



forces of the two fields must have




same magnitude

opposite directions

Electric and Magnetic fields

electric force  = *magnetic force* 

$$eE = Bev$$


$$v = \frac{E}{B} \longrightarrow E = \frac{V}{d} \longrightarrow v = \frac{V}{Bd}$$

Velocity selection

Velocity selector

a device that uses a process of balancing the effects of *electric* and *magnetic* fields to produce a beam of desired *speed*

Velocity selection

Only particles with correct combination of
[*charge* | *mass* | *velocity*] will emerge through slit

Discovering the electron



J.J. Thomson

*the English physicist who
discovered the electron
using the vacuum tube*

J.J. Thomson's summary of observation:

Thomson's conclusions

```
graph TD; A[Thomson's conclusions] --> B[Electrons have negative charge]; A --> C[Electrons remain as a tight beam after deflection]; A --> D[The beam remains straight when there's balance between forces];
```

Electrons have
negative charge

Electrons remain
as a tight beam
after deflection

The beam remains
straight when
there's balance
between forces

J.J. Thomson's summary of observation:

Particles were deflected
towards a **positive** plate

So particles were
negatively charged

From deflection of the
beam by **magnetic field**

J.J. Thomson's summary of observation:

When deflected, the beam
remained as a **tight, single** beam

instead of being **broad** beam

Particles all have same [**mass** | **charge** | **speed**]

J.J. Thomson's summary of observation:

The beam remained **straight** due to the **balance** of the forces of **electric** and **magnetic** field

From there, he could calculate $\frac{e}{m_e}$

Measuring e



Robert Millikan

the American physicist
who measured e

Measuring e

Millikan's procedure



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graph TD; A[Millikan's procedure] --> B[He produced oil droplets using atomizer spray]; A --> C[He determined velocity (and weight) when the electric field was switched off]; A --> D[He determined that the droplet's weight was balanced by the electric force. Switched the field on and adjusted until the droplet remained stationary]; A --> E[The droplet that absorbed an electron would gain negative charge so the electric force on it would change. Included a source of beta-radiation];
```

He produced
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He determined
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He determined
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Switched the field
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The droplet that
absorbed an
electron would gain
negative charge so
the electric force
on it would change

Included a source
of beta-radiation

Measuring e

Millikan found that the charges
were all small multiples of a
particular value, e

Measuring e

So, m_e is now calculated $\frac{e}{m_e}$ and e

Quantization of charge

The electric charge is quantized

which means

The charge must have a
value which is multiple of e